

# In vitro Method to Evaluate the Cleansing Performance of Surfactants Dedicated to Micellar Water Formulations

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## INTRODUCTION

- An easy way to formulate gentle skin cleansers is to reduce the dose of surfactants, as is the case in micellar waters. However this poses the challenge of finding a good compromise with the cleansing properties, especially for makeup removal.

- Surfactant characteristics such as CMC (Critical Micellar Concentration) [1], wetting properties [2], help in comparison but the measurement condition is far from the final formulation combining them with other ingredients **A**. A method closer to the application condition is needed.

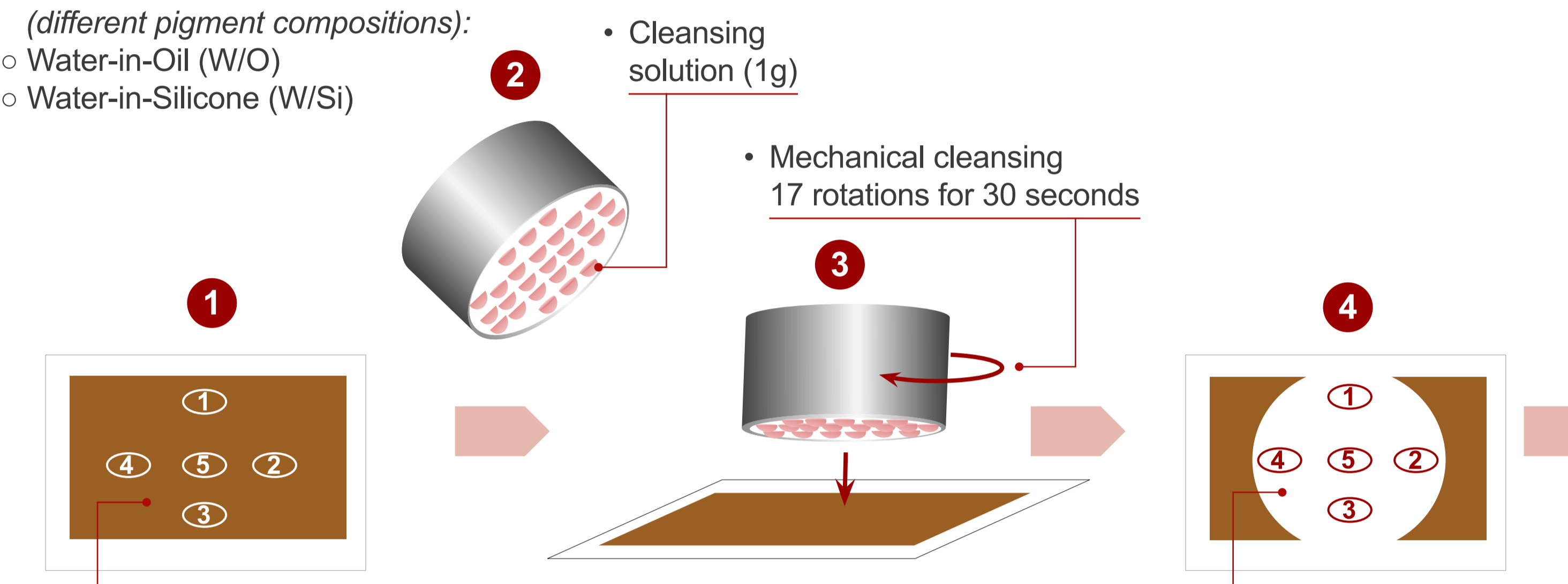
- Objective: develop a simple, quick and reproducible, in vitro method to select effective cleansing ingredients for micellar water and guide the development of the formula.**

## Materials & Methods

### Materials

- Ingredients tested at the same concentration (% AM: Active Matter in demineralized water) alone or in combination.
- Micellar waters benchmark used as such, as in their condition of use.

- Selection of 2 difficult to clean foundations benchmarks (different pigment compositions):
  - Water-in-Oil (W/O)
  - Water-in-Silicone (W/Si)



- Calibrated film of foundation: 30µm thickness (in-house made white carrier)
- L\* parameter (clarity) measurement, after 12h to 24h drying at 40°C / 5 different places of the film (chromameter CR400 - Minolta company)

### Methods:

- Evaluation of make-up removal performance based on colorimetric measurement (L\*, a\*, b\*) on a standard film of foundation before and after mechanical cleansing.
- Protocol adapted to be representative of the final micellar water routine.

Average % of make-up removal

$$\frac{[L^* (\text{with make-up}) - L^* (\text{white control})] - [L^* (\text{after cleansing}) - L^* (\text{white control})]}{[L^* (\text{with make-up}) - L^* (\text{white control})]}$$

- Controls:
- Negative: Cotton alone (rubbing effect)
  - Negative: 1g of hard water (30° french hardness).
  - Positive: Caprylic/ Capric Triglyceride applied on a W/Si benchmark

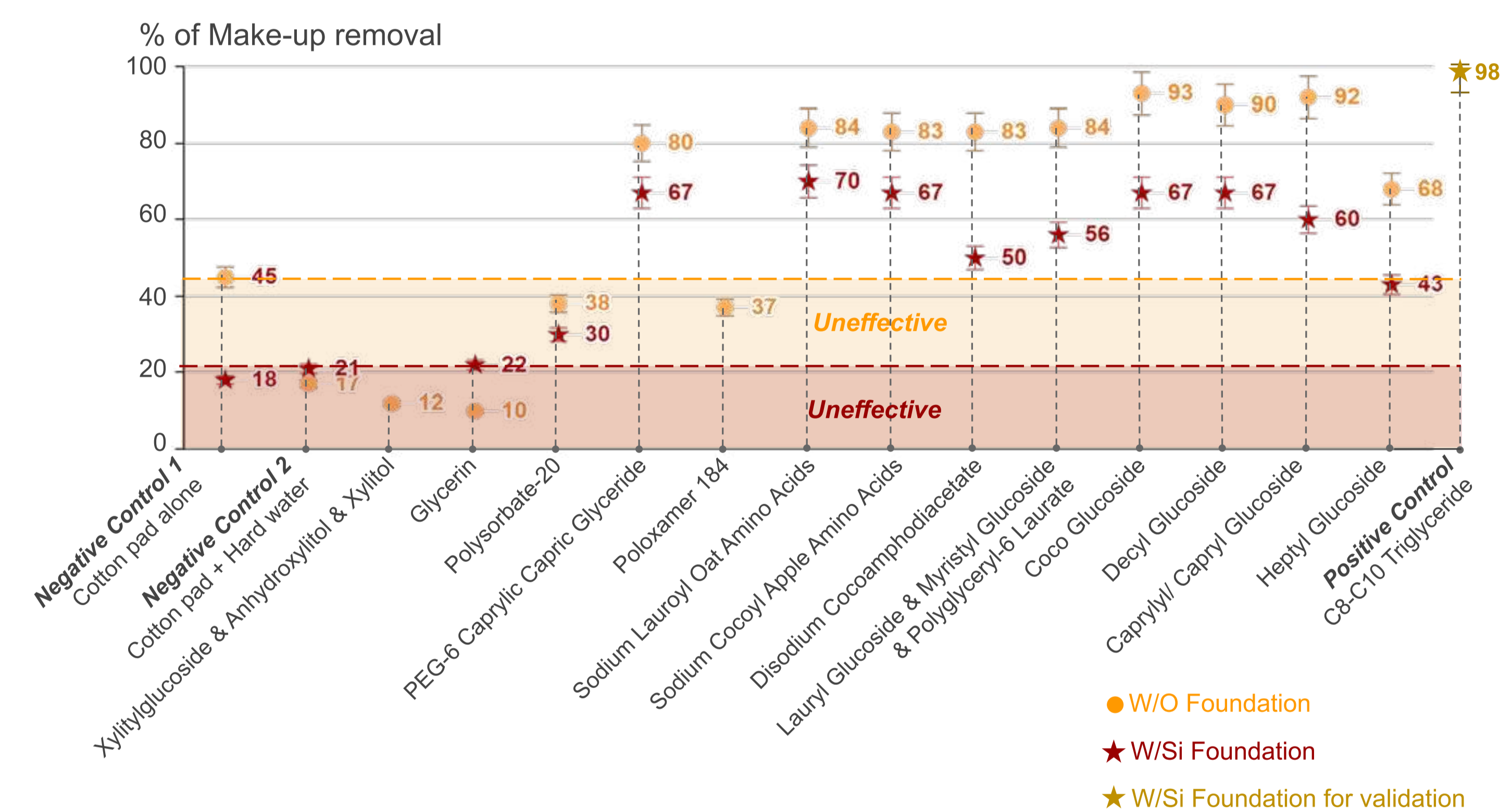
## Results & Discussion

### Method adaptation

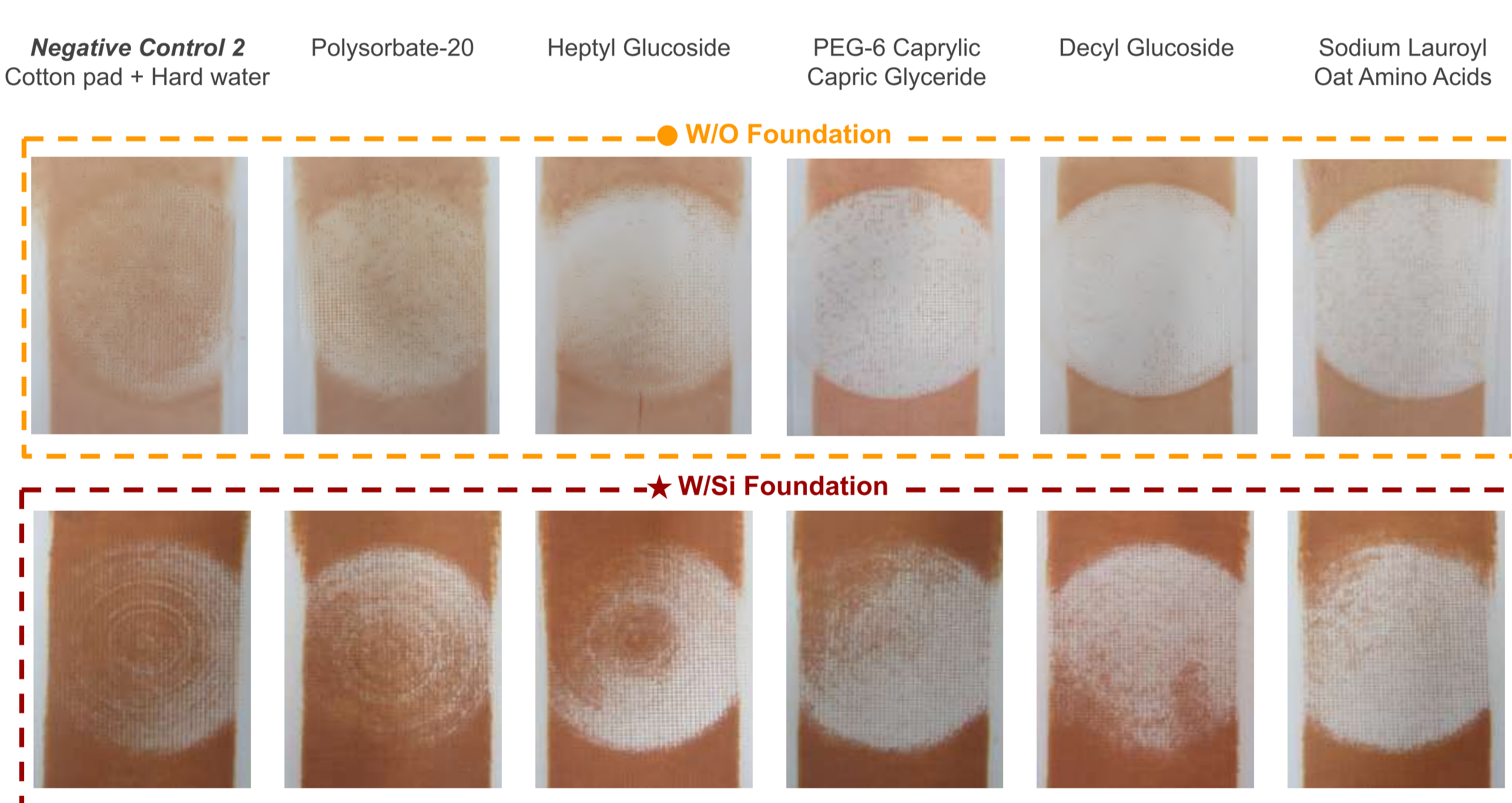
- No effect of the pH of the surfactant solutions (4 to 11) on performance
- Strong impact of the surfactant concentration → 1% AM selected

### Cleansing performance of Ingredients tested alone

Ingredient at 1% Active Matter (AM) in demineralized water - Initial pH



Films after cleansing (Make-up removal with ingredient at 1% Active Matter (AM) in demineralized water - Initial pH)



- Good cleansing properties on both films of foundation: PEG-6 Caprylic Capric Triglyceride, Sodium Cocoyl Apple Amino Acids, Sodium Lauroyl OAT Amino Acids, Coco and Decyl Glucoside.
- No effect of humectant & moisturizing essential additives: Glycerin; Xylitylglucoside complex.
- Ineffective materials: Sodium Cocoyl Glutamate\*, Sodium Lauroyl Sarcosinate\*, Cocamidopropyl Betaine\*, Poloxamer 184, PEG-40 Hydrogenated Castor Oil and Polysorbate-20.
- Effects of Solubilizers: performance of Heptyl glucoside > PEG-40 Hydrogenated Castor Oil = Polysorbate-20.

\*data not shown

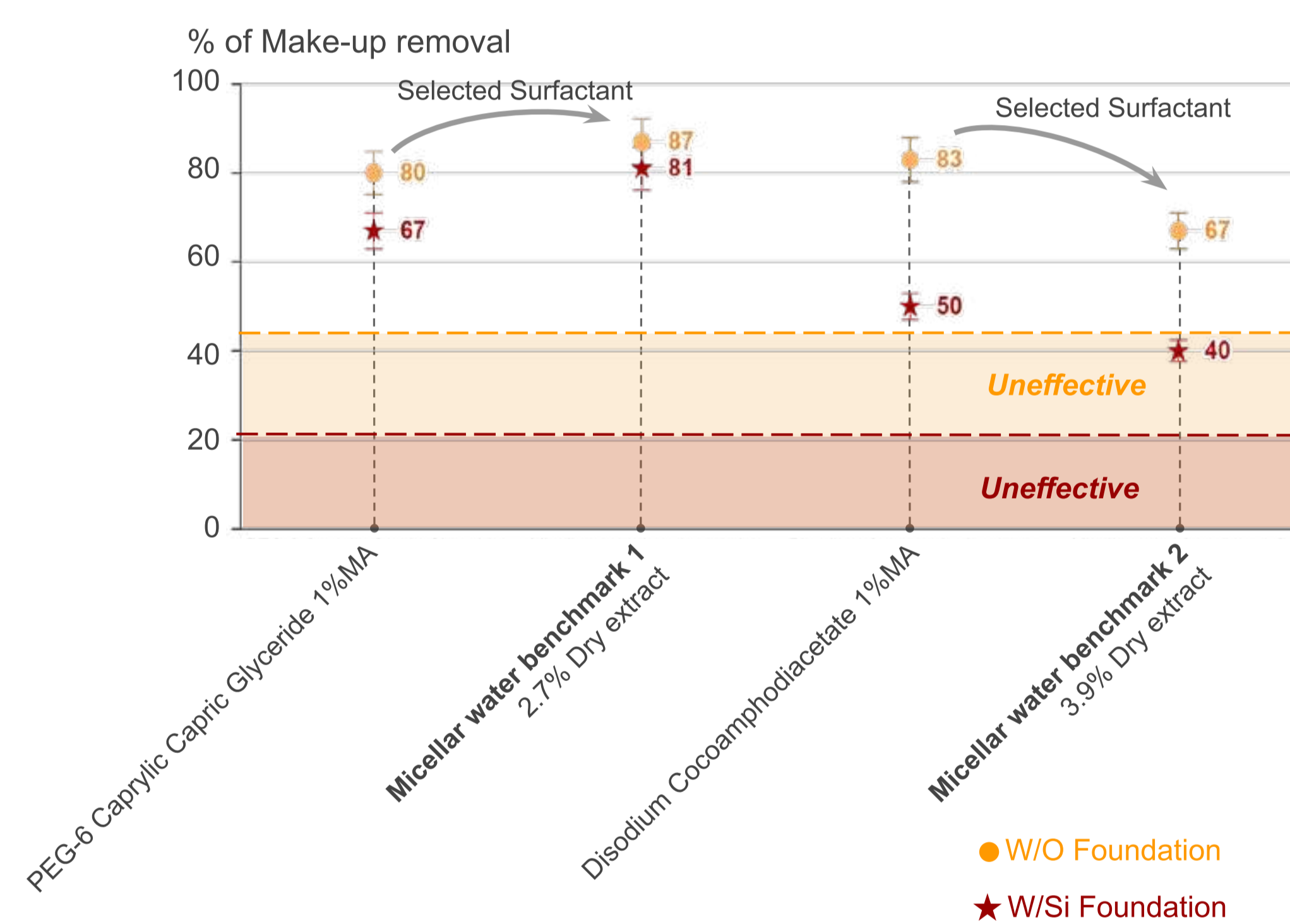
Illustration of discrepancies with physico-chemical parameters

A	CMC (g/L product at 30%AM)	Wetting time (s)	Make-up removal performance
Sodium Lauroyl Sarcosinate*	2.9	25	-- Uneffective
Sodium Lauroyl OAT Amino Acids	2.4	46	++ Good

\*data not shown

### Cleansing performance of micellar water benchmarks

Micellar waters

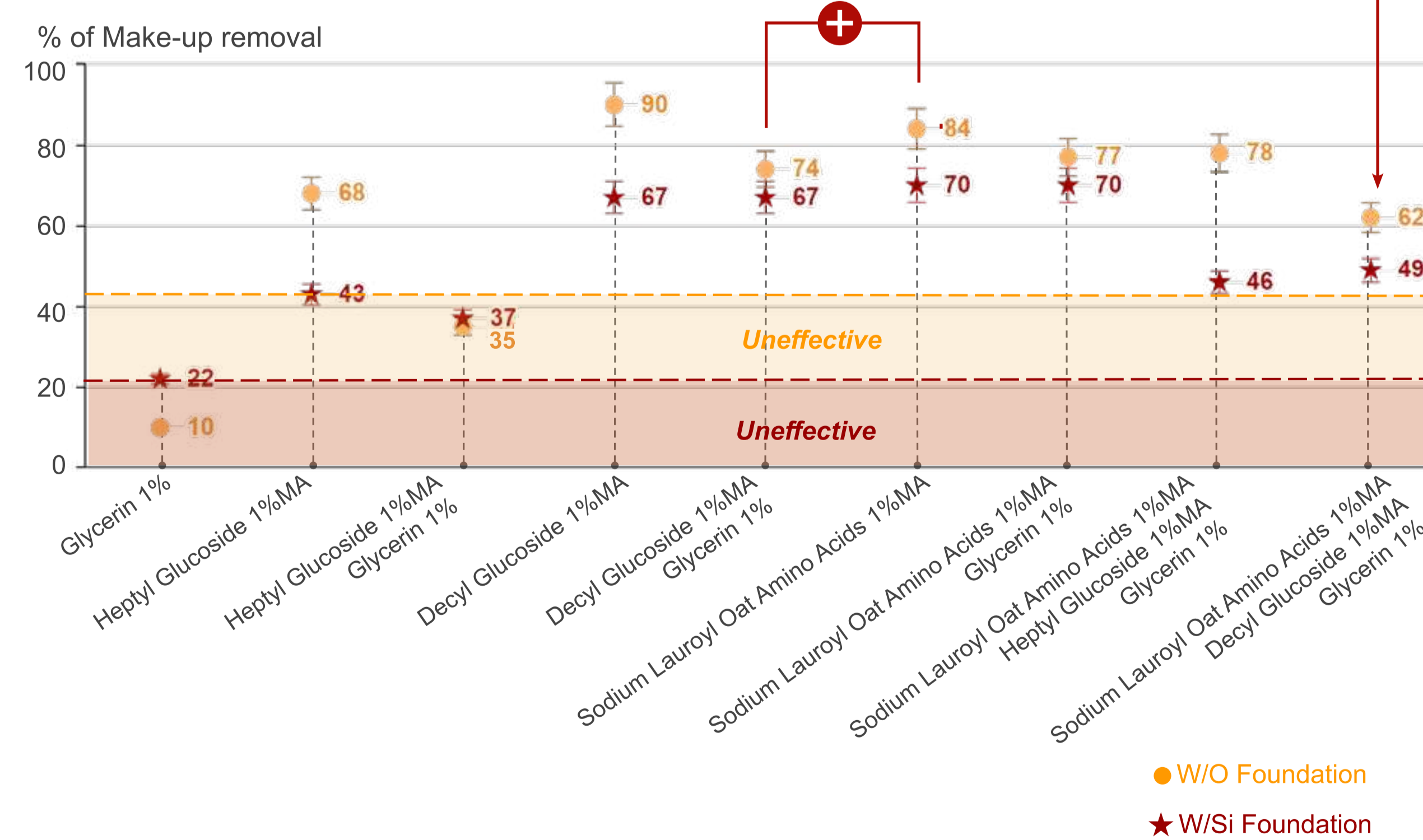


- Suitable method to evaluate finished micellar waters.
- Same reaction pattern between micellar water and their respective surfactant regarding versatile effect/ or not on the 2 foundations.
- Noticeable impact of other ingredients on the cleansing.

### Cleansing performance of Ingredient combinations

Combinaison of ingredients in demineralized water - Initial pH

No demonstrated interest in surfactant combinations despite double active matter content.



- Confirmed impact of other ingredients than surfactants, depending on surfactant nature.
- Water-in-Silicone foundation overall harder to remove on the 2 foundations.
- Noticeable impact of other ingredients on the cleansing.

## References

- Rolls, E. (2020). Clean Chemistry—The Science Behind Cleansing Products. *Discovering Cosmetic Science*, 19.
- Noh H, Kang T, Ryu JS, Kim SY, Oh S-G (2016) Synergy effect for performance of anionic SDS/ADS mixtures with amphoteric and nonionic surfactants. *J. of Korean Oil Chemists' Soc.* 33; 3: 449-45.

## CONCLUSION

- Development of a simple & reproducible in vitro test (maximum standard deviation of 6 %) to screen the make-up removal performance of surfactant solutions, at a concentration level similar to those used in micellar waters.
- Method also suitable for studying the influence of ingredient combinations to optimize micellar water formulations.

